	Utech
Name:	
Roll No.:	To Alarma Of Exercising 2nd Explored
Invigilator's Signature :	

CS/B.TECH (CT)/SEM-7/CT-703C/2011-12 2011

OXIDE CERAMICS

Time Allotted: 3 Hours Full Marks: 70

The figures in the margin indicate full marks.

Candidates are required to give their answers in their own words as far as practicable.

Answer any *five* questions. $5 \times 14 = 70$

- 1. Why ZrO_2 and ZrO_2 -bearing oxides are used in metallurgical and high temperature Chemical Engineering industries? What are the structures exhibit of ZrO_2 and in what temperature they are stable? In what process toughened ceramics are developed? Write in short the properties of toughened ceramics. 2+3+4+5
- 2. Write shot notes on the following : $4 \times 3\frac{1}{2}$
 - a) Tin dioxide
 - b) Berillia
 - c) Thoria
 - d) Mechanical Properties of Pure Oxide.

7503 [Turn over

- 3. What do you mean by pure oxide? How pure oxides are classified? State some important properties of Pure oxide? Write in short the general methods of fabrication of pure oxide body?
 2 + 2 + 4 + 6
- 4. How is ZrO_2 partially stabilized and why? What are the application of Zirconia Ceramics? 3 + 4 + 7
- 5. Name different polymorphic form of alumina available in nature. Discuss briefly the structure of any one form. What are the differences between reactive and non-reactive alumina? Discuss briefly the application of both form of aluminas in ceramic products. 2 + 4 + 4 + 4
- 6. Define magnesium aluminate spinel. Discuss its structure. Discuss briefly about pure phase spinel bodies preparation in the laboratory for different advanced applications.

3 + 3 + 8

7. Why is DBM produced from sea water contain ${\rm SiO}_2$? How is ${\rm SiO}_2$ present in DBM detrimental in refractories? Discuss the structure of MgO. Explain how can magnesia bricks having following properties be produced in the plant?

```
MgO — 98 wt% ( min )
```

SiO
$$_2$$
 — 0·1 wt% (max)

B.D.
$$-3.00 \text{ gms/cc}$$
 (min)

RUL (ta)
$$-$$
 + 1700°C (min.) $2 + 3 + 3 + 6$

7503 2