						(Unexan		
Nar	ne : .	• • • • • • •		•••••	• • • • •			
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Invi	igilato	r's S	ignature :	•••••				
			CS/B.TE	CH (CHE)	/SI	EM-4/CHE-403/2010		
				2010				
	MA	ATE	RIALS SCI	ENCE AN	ID	TECHNOLOGY		
Time Allotted: 3 Hours						Full Marks : 70		
		Th	ne figures in the	e margin ind	dica	ate full marks.		
Car	ndida	tes ai		ive their an as practica		ers in their own words as		
				GROUP – A	_			
			( Multiple C	hoice Type	gı	uestions )		
1.	Choose the correct alternatives for any <i>ten</i> of the following :							
						$10 \times 1 = 10$		
<ul><li>i) The tiny block formed by the a of atoms is called</li></ul>					ar	rangment of small group		
		a)	unit cell	1	b)	space lattice		
		c)	lattice point		d)	none of these.		
	ii) According to Bravais, there are possibly types of space lattice in seven basic crystal system.							
		a)	8	1	b)	14		
		c)	20	•	d)	24.		
	iii)	le cube is approximately						
		a)	0.4	1	b)	0.42		
		c)	0.52		d)	0.8.		

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iv)	Rec	overy process in cold	worke	ed metal can be studied				
	by Same of the sam							
	a)	hardness	b)	resistivity				
	c)	fracture toughness	d)	Young's modulus.				
v)		ical resolved shear sulated by applying	stress	in a single crystal is				
	a)	Brag's law	b)	Hook's law				
	c)	Schmid's law	d)	Frank-Reed law.				
vi)	Perlite is obtained when steel is							
	a)	quenched in oil						
	b)	cooled in still in air						
	c) slowly cooled in furnace							
	d)	quenched in water.						
vii)	Flux is added to ore in the smelting operation to							
	a) increase melting point of slag							
	b) decrease melting point of slag							
	c) increase melting point of gangues							
	d) decrease melting point of ore.							
viii)	Sinter roasting is done for the ores							
	a) that are found in fine form							
	b) that are free flowing							
	c) that contains high % of gangues							
	d) that contains less % of gangues.							



- Predominant area diagram helps in determining ix)
  - partial pressure of oxygen at which a metal oxide can be reduced to metal
  - b) partial pressure of sulphur dioxide at which a metal sulphide can be reduced to metal
  - stable form of any metal compound at a specific c) % of SO  $_2$  & O  $_2$
  - all of these. d)
- Iron obtained from blast furnace is known as X)
  - a) wrought iron
- b) cast iron
- pig iron c)
- d) soft iron.
- Cowper stoves are provided in a blast furnace to xi)
  - a) provide hot air supply to it
  - supply coke to charge b)
  - provide a stove in which coal is burned to get coke c)
  - none of these. d)
- xii) Leaching is a unit operation associated with
  - hydrometallurgy a)
- b) electrometallurgy
- c) pyrometallurgy
- d) refining.

### GROUP - B

#### (Short Answer Type Questions)

Answer any *three* from the following.  $3 \times 5 = 15$ 

2. Establish the relationship 
$$d = \frac{1}{\left(\frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2}\right)}$$
 for the

distance between adjacent planes in a crystal.



- 3. Write short notes on *T-T-T* diagram.
- 4. Why is smelting required? What are the differences between reduction smelting and metallothermic smelting? Discuss the reduction smelting of iron ore. 2+3
- 5. Differentiate between cold and hot working of metals. 5
- 6. Differentiate between hydometallurgy and pyrometallurgy. Show the steps in the Electrometallurgical process of extraction of aluminium. 2+3

#### GROUP - C

## (Long Answer Type Questions)

Answer any *three* of the following.  $3 \times 15 = 45$ 

- 7. a) What do you mean by powder metallurgy?
  - b) State the advantages and limitations of powder metallurgy. 3 + 12
- 8. a) What is hardening? Describe the process briefly.
  - b) What is martempering?

(4+6)+5

- 9. Explain the property and microstructural changes during cold working and annealing of metals.
- 10. a) What is corrosion ? Explain different methods to prevent corrosion. 1+5
  - b) Define plastic deformation and strain hardening. Explain the mechanism of slip. (1+5)+2+2+5
- 11. a) Why does the fatigue fracture occur? Describe the ways by which fatigue life can be improved.
  - b) What is galvanic corrosion and how is it protected ? What is the role of non-metallic coating on corrosion prevention? (2+2)+6+5

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